

GLYCEROKINASE (GlyK)

[EC 2.7.1.30]

from recombinant *E. coli*



SPECIFICATION

State : Lyophilized
Specific activity : more than 80 U/mg protein

PROPERTIES

Subunit molecular weight : ca. 54,700
Optimum pH : 9.8 (Fig. 1)
pH stability : 5.5 - 9.0 (Fig. 2)
Thermal stability : No detectable decrease in activity up to 40 °C. (Fig. 3)
Optimum temperature : above 50 °C (Fig. 4)
Michaelis constants : (186 mM Glycine-Hydrazine-KOH buffer pH 9.8, at 30 °C)
Glycerol 0.026 mM
ATP 0.025 mM

STORAGE

Stable at -20 °C for at least six months

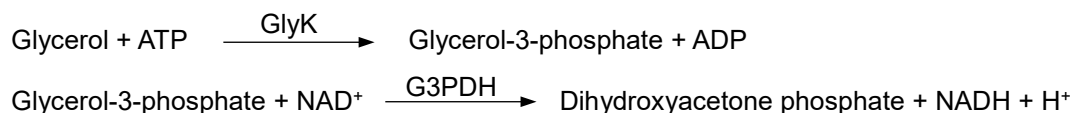
APPLICATION

The enzyme is useful for enzymatic determination of glycerol and triglyceride when coupled with glycerol-3-phosphate dehydrogenase

ASSAY

Principle

The change in absorbance is measured at 340 nm according to the following reactions.



Unit Definition

One unit of activity is defined as the amount of GlyK that forms 1 μmol of NADH per minute at 30 °C.

Solutions

- I Buffer solution ; 200 mM Glycine-Hydrazine-KOH, pH 9.8 (Dissolve 1.5 g glycine and 5 mL hydrazine hydrate in 80 mL distilled water. After adjusting pH to 9.8 with 1 M KOH, fill up to 100 mL with distilled water.)
- II MgCl_2 solution ; 100 mM (2.03 g $\text{MgCl}_2 \cdot 6\text{H}_2\text{O}$ /100 mL distilled water)
- III ATP solution ; 100 mM (0.605 g ATP disodium salt $\cdot 3\text{H}_2\text{O}$ /(8.2 mL distilled water + 1.8 mL 1 M NaOH))
- IV NAD^+ solution; 100 mM (0.663 g NAD^+ free acid/10 mL distilled water)
- V Glycerol-3-phosphate dehydrogenase ; 1700 U/mL (from rabbit muscle, Roche Diagnostics)
- VI Glycerol solution ; 330 mM (3.04 g Glycerol/100 mL distilled water)

Preparation of Enzyme Solution

Dissolve the lyophilized enzyme with distilled water and dilute the enzyme solution to 0.1 to 1.0 U/mL with 50 mM Tris-HCl buffer pH 9.0 containing 0.1 % bovine serum albumin.

Procedure

1. Prepare the following reaction mixture and pipette 2.70 mL of reaction mixture into a cuvette.

Solution I	27.90 mL	Solution IV	0.15 mL
Solution II	0.57 mL	Solution V	0.30 mL
Solution III	0.39 mL	Solution VI	0.30 mL
Distilled water	0.39 mL		
2. Incubate at 30 °C for about 3 minutes.
3. Add 0.015 mL of enzyme solution into the cuvette and mix.
4. Read absorbance change at 340 nm per minute (ΔAbs_{340}) in the linear portion of curve.

Calculation

$$\text{Volume activity (U/mL)} = \frac{(\Delta\text{Abs}_{340}) \times (2.70 + 0.015)}{6.22 \times 0.015} \times \text{d.f.}$$

$$\text{Specific activity (U/mg protein)} = \frac{\text{Volume activity (U/mL)}}{\text{Protein concentration (mg/mL)}^*}$$

d.f. ; dilution factor

6.22 ; millimolar extinction coefficient of NADH ($\text{cm}^2/\mu\text{mol}$)

*Protein concentration ; determined by the absorbance at 280 nm (Abs_{280}), where 1 Abs_{280} = 1 mg/mL

REFERENCE

1. Mike J. Comer, Chris J. Bruton, and Tony Atkinson ; *J. App. Biochem.* **1**, 259-270 (1979)

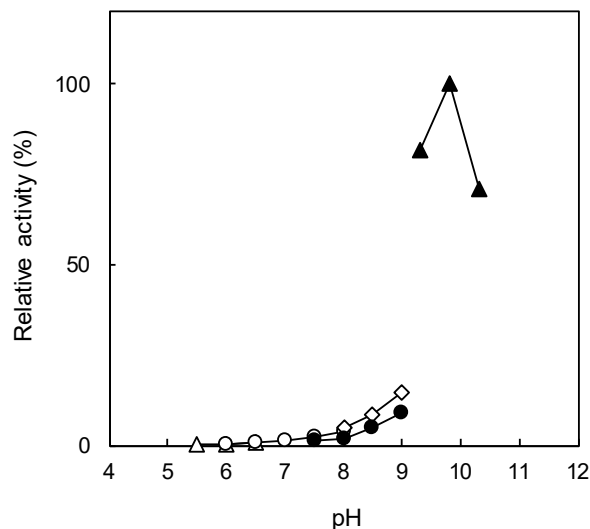


Fig. 1 pH profile

(\triangle MES, \circ phosphate,
 \bullet Tris-HCl, \diamond Bicine,
 \blacktriangle Glycine-Hydrazine)

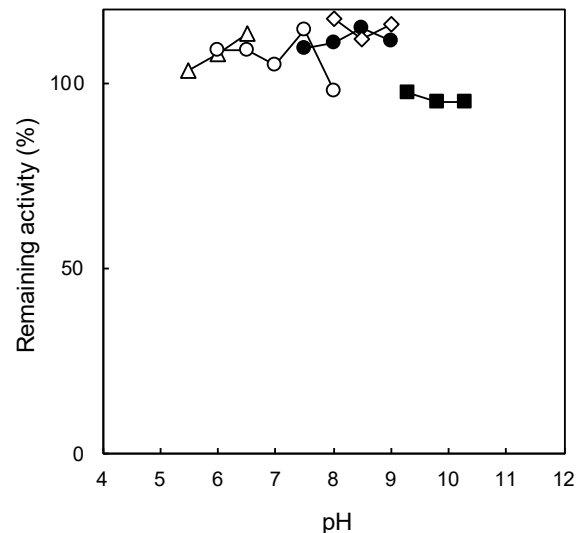


Fig. 2 pH stability

(treated for 24 hr at 4 °C in the
 following buffer solution (0.1 M);
 \triangle MES, \circ phosphate,
 \bullet Tris-HCl, \blacksquare Glycine-
 Hydrazine, \diamond Bicine)

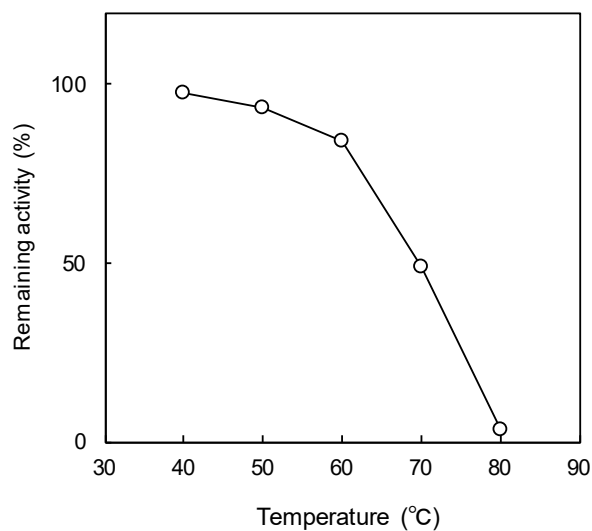


Fig. 3 Thermal stability

(treated for 15 min in 100 mM
 Bicine buffer, pH 8.5)

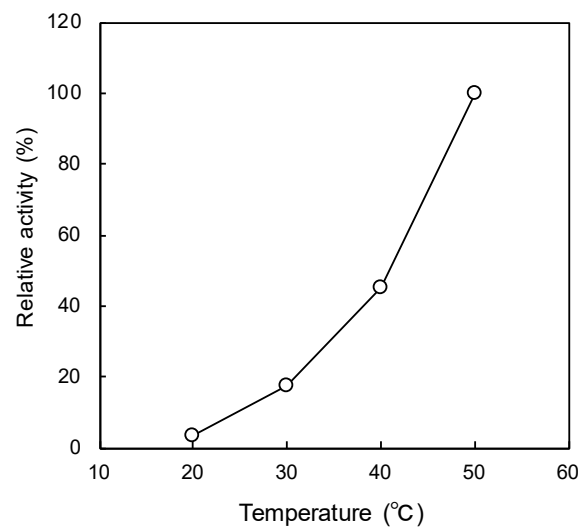


Fig. 4 Thermal activity