

# α-GLUCOSIDASE (α-Glu)

[EC 3.2.1.20]

from Bacillus stearothermophilus

 $\alpha$ -D-Glucoside + H<sub>2</sub>O  $\leftrightarrow$  D-Glucose + Alcohol

# **SPECIFICATION**

State : Lyophilized

Specific activity : more than 40 U/mg protein Contaminants : (as  $\alpha$ -Glu activity = 100 %)

Phosphoglucomutase < 0.01 % NADH oxidase < 0.01 % Alcohol dehydrogenase < 0.01 %

# **PROPERTIES**

Molecular weight : ca. 50,000

Optimum pH : 6.0 - 7.0 (Fig. 1) pH stability : 5.0 - 11.0 (Fig. 2)

Isoelectric point

Thermal stability : No detectable decrease in activity up to 60 °C. (Fig. 3, 4)

Michaelis constants : (50 mM Potassium phosphate buffer, pH 6.3, at 30 °C) p-Nitrophenyl-α-glucopyranoside (PNPG) 0.73

p-Nitrophenyl-α-glucopyranoside (PNPG) 0.73 mM Maltose 1.3 mM Phenyl-α-glucopyranoside 2.4 mM

Substrate specificity : PNPG 100 %

Maltose 177 % Phenyl-α-glucopyranoside 59 %

# **STORAGE**

Stable at -20 °C for at least one year

### **APPLICATION**

The enzyme is useful for diagnostic reagent, for example,  $\alpha$ -amylase determination.



### **ASSAY**

# **Principle**

The change in absorbance is measured at 400 nm according to the following reaction.

p-Nitrophenyl-
$$\alpha$$
-glucopyranoside (PNPG)  $\xrightarrow{\alpha$ -Glu p-Nitrophenol (PNP) + Glucose

### **Unit Definition**

One unit of activity is defined as the amount of  $\alpha$ -Glu that forms 1  $\mu$ mol of PNP per minute at 30 °C.

# **Solutions**

- I Buffer solution; 100 mM Potassium phosphate buffer, pH 6.3
- II PNPG solution; 20 mM (0.603 g PNPG/100 mL distilled water) (Stable for two weeks if stored at 0 5 °C)
- III Na<sub>2</sub>CO<sub>3</sub> solution; 0.2 M (2.12 g Na<sub>2</sub>CO<sub>3</sub>/100 mL distilled water)

# **Preparation of Enzyme Solution**

Dissolve the lyophilized enzyme with distilled water and dilute to 0.006 to 0.022 U/mL with 10 mM Potassium phosphate buffer containing 1 mg/mL BSA, pH 7.5.

# **Procedure**

1. Prepare the following reaction mixture and pipette 1.5 mL of reaction mixture into a test tube.

Solution I 10.0mL

Solution II 5.0mL

- 2. Incubate at 30 °C for 5 minutes.
- 3. Add 0.5 mL of the enzyme solution and mix.
- 4. Incubate at 30 °C for exactly 15 minutes.
- 5. After incubation, add 2.0 mL of Solution III and mix.
- 6. Read absorbance at 400 nm (Abs•test).

At the same time, prepare the blank with 1.5 mL of the reaction mixture, and add 2.0 mL of Solution III after incubation at 30 °C for 15 minutes, followed by addition of the enzyme solution (Abs•blank).

### Calculation

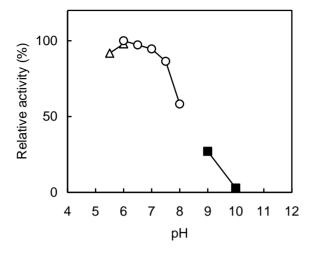
Volume activity (U/mL) = ((Abs•test) - (Abs•blank)) 
$$X = \frac{4.0}{18.1 \times 15 \times 0.5} \times d.f.$$

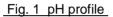
Specific activity (U/mg protein) = 
$$\frac{\text{Volume activity (U/mL)}}{\text{Protein concentration (mg/mL)}^*}$$

d.f.; dilution factor

18.1; millimolar extinction coefficient of PNP (cm²/µmol) \*Protein concentration; determined by Bradford's method







△ acetate,O phosphate,Gly-NaOH

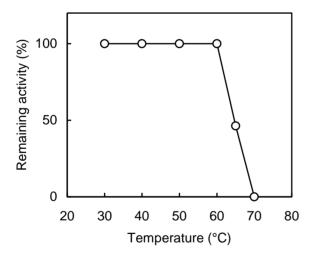


Fig. 3 Thermal stability

treated for 15 min in 0.1M potassium phosphate buffer, pH 8.0

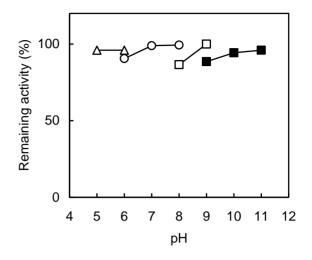


Fig. 2 pH stability

treated for 24 hr at 4 °C in the following buffer solution (0.1 M); △ acetate, O phosphate, ☐ TEA-NaOH, ■ Gly-NaOH

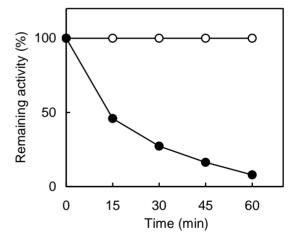


Fig. 4 Thermal stability

treated for in 0.1M potassium phosphate buffer, pH 8.0 O 60 °C, ● 65°C